

**Chemistry 108**  
Fundamentals of Chemistry  
Spring 2010

**Course Instructor:** Professor Stephen Dunham  
Office: 214 Collier Hall of Science, 610-861-7103  
Email: [stephendunham@moravian.edu](mailto:stephendunham@moravian.edu)

**Lecture:** M,W, F 10:20-11:10, 204 Collier  
**Problem Session:** R 10:20-11:10, 204 Collier

**Office Hrs:** Posted each week on **blackboard**, or by appointment.

**Lab Instructor:** Professor Carol Libby  
Office: 213 Collier Hall of Science, 610-861-1629  
Email: [cplibby@cs.moravian.edu](mailto:cplibby@cs.moravian.edu)

**Labs:** M or F 1:15-4:15, 211 Collier  
T 8:30-11:30, 211 Collier

**Required Materials:**

**Text:** Essentials of General, Organic, and Biochemistry An Integrated Approach, by Denise Guinn, Rebecca Brewer, 2010, W.H. Freeman.

**Scientific Calculator:** Must be able to calculate logs | no cell phones!

**Homework-Journal:** This must be a bound notebook with numbered pages. You will be allowed to use this book for all quizzes and exams.

**Black Board Web Site:** <http://blackboard.moravian.edu/>

**You must enroll in our Chemistry 108 blackboard site.**

Throughout the semester, all handouts will be posted to the course blackboard page. Please access this early and often!

**Safety glasses:** Will be provided in the laboratory. They must be worn at all times!

**Course Goals:**

- Use real-life examples, particularly those that are health related, to illustrate the relationship between chemical principles and living organisms.
- Relate the properties of atoms and molecules with the organization of elements in the periodic table.
- Recognize relationships between physical properties of atoms, compounds, and molecules; and the physical states of matter, solubility, reactivity, molecular shape, and biochemical function.
- Apply qualitative and quantitative aspects of chemistry to problem solving.
- Use the scientific method to actively seek knowledge through the study of chemical processes in a controlled environment.

**Attendance:** It will be *very difficult* for you to learn chemistry concepts and follow them over the semester if you miss course meetings (class, problem sessions, and laboratories). As a reminder, the college policy on attendance can be found at <http://www.moravian.edu/studentLife/handbook/academic/academic.html>. If you anticipate an unavoidable absence, please *notify me ASAP before* you are absent. Makeup quizzes, exams, and labs are given at the discretion of the instructor.

**Academic Honesty:** Please be familiar with the college policy on academic honesty <http://www.moravian.edu/studentLife/handbook/academic/academic2.html>. Because this course involves small group learning activities, each student may exchange experimental details and data with her/his lab partner and classmates. However, any work submitted in your name is to be your work alone. You are encouraged to discuss work with others on assignments and labs, but merely copying answers is not acceptable.

**Learning Differences:** Students should contact the Office of Learning Services for disclosure of a learning difference and to request appropriate amendments to this course <http://www.moravian.edu/studentLife/handbook/academic/academic4.html>.

**Grading:** You are not in competition with anyone else in this class. Your grade will be determined only by the percentage of the total points you achieve. In the event that the class average on any exam, quiz, or lab falls below 75%, that score will be curved so that the class average is 75%.

**Percentage Based Grading Scale**

97-100	A !	73-76	C
93-96	A	70-72	C-
90-92	A-	67-69	D+
87-89	B+	63-66	D
83-86	B	60-62	D-
80-82	B-	< 60	F
77-79	C+		

There are a total of 1000 pts that will be factored into your final grade in this course. A point on an exam, quiz, or laboratory counts the same amount.

Exams	550 pts
Quizzes	150 pts
Participation	50 pts
<u>Labs</u>	<u>250 pts</u>
Total	1000 pts

**Exams:** Three in-class exams (130 pts each) and one final (160 pts).  
*You can use your homework-journal as a resource during the exam and quizzes.*

While the material you will be responsible for on each exam could vary, (dependent upon the pace of the course) the following dates will be used for examinations:

Exam 1	Friday, February 12
Exam 2	Friday, March 19
Exam 3	Friday, April 16
Final	Thursday, May 6

**Quizzes:** During the first lecture of each week, a quiz (15 pts each, 12 quizzes per semester) will be given on previous weeks lecture material. Your lowest two quiz scores will be dropped at the end of the term.

**Participation:** Classroom activities require input from everyone in a group. Your group work will result in a mark (+, 0, or -) for each activity. At the end of the term these marks will be converted into a participation score out of 50 pts.

**Labs:** Twelve laboratory experiences (25 pts for each lab) consisting of:

- A group write up that is handed in before leaving lab 18pts
- Group participation, being on time, and working safely 3pts
- Review questions about the laboratory on blackboard 4pts

Your two lowest lab scores will be dropped at the end of the term.

**Makeup Quizzes, Labs, and Exams:** Missed quizzes and labs will typically be counted as one of the “dropped” scores (see grading section above). You are responsible for understanding the content of the material covered during a missed quiz or lab. Makeup exams will be given at the discretion of the instructor for absences that have been documented by the Dean of Students Office and/or a health professional.

**Homework-Journals:** Activities and homework problems will be given for each class meeting. These will not be graded, and are intended to provide you with the “minimum” exercises that cover the important concepts in that chapter. The content for quiz and exam questions often comes directly from these activities and homework problems.

- Only problems from *Guinn and Brewer*, or class activities can be in your journal
- No loose papers or sticky notes are allowed
- Use **ink** for all journal writing. If you make a mistake, draw a single **X** or line through error. Do not scribble or scratch out your writing.
- Journals can be used as a resource for exams and quizzes, and **must be handed in with each exam**. Extra information, missing pages, or scratched out words will result in lost points for that exam.

**Class Participation:** Nearly all concepts in this course will build upon each other, and this requires you to understand the material in previous activities to build a bridge to the new material we will be learning.

If you have questions that are not answered in class, you can come to my office hours, schedule an appointment, or start a blog (public) or journal (private) entry on the **Blackboard** web site.

**Class Etiquette:**

- Turn off or silence cell phones! NO-text messaging during class
- Do not record or take pictures of classmates or instructors without their permission

**Email Etiquette:** Although email may seem like an instantaneous form of communication, it is not. Just because you sent me an email, does not mean that I have: 1) read it, 2) understood it, and/or 3) approved any requests you made in it.

- I will reply individually, or as a class response to all email received.
- Assume that email sent between the hours of 9 PM and 9AM has NOT been read

**Tips for Success:**

1. The course uses a group learning format that depends upon your **active** contribution on a daily basis. **You will be asked to explore data and make predictions BEFORE any class discussion has taken place.** This process is unconventional for many science courses, and it may be uncomfortable at first because you will need to rely on logic and not on “finding the answer” by *Googling* your textbook or lecture notes.
2. After class, read the text that compliments the activity, and work the suggested problems *in your homework-journal*. If you are struggling with the problems, redo the activity on your own or in a small group to reinforce the concepts.
3. If you are studying in a group, be certain that you can **“go it alone”**. Spend some time alone answering the same problems, or attempting related problems in the text.

**Pace of the Course:** The schedule below is a guide for course coverage this semester.

Week	Beginning		Anticipated Text Coverage
Jan	18	Ch # 1	Measurement, Atoms, and Elements
Jan	25	Ch # 2	Compounds
Feb	1	Ch # 3	Shapes of Molecules and Their Interactions
Feb	8	Ch # 3-4	<b><u>Exam #1 CH 1-3</u></b>
Feb	15	Ch # 4	Solids, Liquids, and Gases
Feb	22	Ch # 5	Solutions, Colloids, and Membranes
Mar	1	Ch # 6	Hydrocarbons and Structure
Mar	8	<b><i>SPRING BREAK</i></b>	
Mar	15	Ch # 7	Organic Functional Groups <b><u>Exam #2, CH 4-7</u></b>
Mar	22	Ch # 8	Chemical Reaction Basics
Mar	29	Ch # 9	Acids, Bases, pH and Buffers
April	5	Ch # 11	Proteins: Structure and Function
April	12	Ch # 12	Carbohydrates: <b><u>Exam #3, CH 8-11</u></b>
April	19	Ch # 13	Lipids: Structure and Function
April	26	Ch # 15	Nucleic Acids: DNA and RNA
May	3	<b>Final Exam, May 6, 8:30 AM <u>CH 12-15, and Cumulative CH 1-11</u></b>	